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Studies on Disaccharide Nucleoside Synthesis. Mechanism of the Formation of Trisaccharide Purine Nucleosides

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STUDIES ON DISACCHARIDE NUCLEOSIDE SYNTHESIS. MECHANISM OF THE FORMATION OF TRISACCHARIDE PURINE NUCLEOSIDES

Sergey N. Mikhailov^{a,*}, Andrei A. Rodionov^a, Ekaterina V. Efimtseva^a, Boris S. Ermolinsky, Marina V. Fomitcheva^a, Nelly Sh. Padyukova^a, Klaus Rothenbacher^b, Eveline Lescrinier^b, Piet Herdewijn^b

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ABSTRACT: The unexpected formation of trisaccharide nucleosides during synthesis of purine 5'-O- β -D-ribofuranosylnucleosides in the presence of Lewis acids was observed.

Purine nucleosides having a D-ribofuranosyl substituent bonded to the 2'-hydroxyl function of the nucleoside sugar moiety have been isolated from tRNA¹. Recently we have developed general route for the preparation of 2'-O- β -D-ribofuranosylnucleosides by condensation of partially protected nucleosides with 1-O-acetyl-2,3,5-tri-O-benzoyl- β -D-ribofuranose (1) in the presence of tin tetrachloride². This reaction was further extended to the synthesis of pyrimidine 5'-O- β -D-ribofuranosylnucleosides³.

The application of this reaction to the preparation of purine 5'-O-β-D-ribofuranosylnucleosides gave a complex mixture of products⁴. Thus condensation of 2 with slight excess of 1 in the presence of 1.2 eq. of SnCl₄ in dichloroethane (0°C, 2 hr, under nitrogen) produced a mixture of compounds 3, 4 and 5 which was separated on silica gel column (yields 18%, 29% and 17% respectively). Further deprotection of 3, 4 and 5 with ammonia in methanol gave adenosine, 6 and 7. The structures of 4-7 were proven by NMR and mass spectroscopy. The formation of trisaccharide 7 as a main by-product was evidently due to the instability of glycosidic bond of adenine nucleosides in the presence of

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Lewis acid. Nucleoside 2 was treated with tin tetrachloride for 16 h at 0° C, after usual work-up and saponification, the mixture of adenine, adenosine and 6 was separated by HPLC. It should be mentioned that the formation of disaccharide in this case without acceptor of nucleic base 1 is much slower.

It may be concluded that purine nucleoside is the source of additional ribose moiety in 7. To exclude the formation of by-products another substrate with more stable glycosidic bond should be used. The alternative route to 6 was proposed starting from 2',3'-O-isopropylidenenucleosides.

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REFERENCES

- 1. Limbach, P.A.; Crain, P.F.; McCloskey, J.A. Nucleic Acids Res., 1994, 22, 2183-2196.
- 2. Mikhailov, S.N.; Efimtseva, E.V.; Gurskaya, G.V.; Zavodnik, V.E.; De Bruyn, A.; Rozenski, J.; Herdewijn, P. J. Carbohydr. Chem., 1997, 16, 75-92.
- 3. Mikhailov, S.N.; Rodionov, A.A.; Efimtseva, E.V.; Fomitcheva, M.V.; Padyukova, N.Sh.; Herdewijn, P.; Oivanen, M. Carbohydr. Lett. 1997, 2, 321-328.
- Mikhailov, S.N.; Rodionov, A.A.; Efimtseva, E.V.; Ermolinsky, B.S.; Fomitcheva, M.V.; Padyukova, N.Sh.; Rothenbacher, K., Lescrinier, E.; Herdewijn, P. Eur. J. Org. Chem., 1998, in press.